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In-class exercise

1. In an aerobic environment, which is most common, Fe^{3+} or Fe^{2+} ?

2. Which of the following is the most likely geometry for iron in the oxidation state you chose?



3. Of the ligands, N, O and S, which is most favored for (1) according to HSAB?

4. How can you use the chelate effect to further increase the stability of a complex of (1)?

5. You've just designed a siderophore! Sketch a cartoon of what it might look like.